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## ACS PHYSICAL CHEMISTRY THERMOCHEMISTRY STUDY GUIDE

- Which direction does the reaction proceed if  $G$  is positive?: To the left / to reactants
- Which direction does the reaction proceed if  $G$  is negative?: To the right/ to products
- How to predict  $S(\text{system})$ ? (change in entropy of a system): Positive  $S(\text{system})$  means available microstates increases/disorder increases
- How to calculate the  $S_o$  (aka standard entropy change) of the formation of one mole of gas from its elements?: 1. Write the balanced chemical equation with the proper coefficients that make the product ONE MOLE.

2. Find the value of the one mole of product in Appendix B. This is the  $\Delta S(\text{standard})$  for the product.

3. Find the value of the reactants in Appendix B for  $\Delta S(\text{standard})$  and multiply them by the coefficients you got in step 1.

4. Use  $\text{total} = \text{products} - \text{reactants}$  for your final value.

- How to calculate  $G$  (free energy) using enthalpy  $H$  and entropy  $S$  values?-

$$: \Delta G = \Delta H - T\Delta S$$

- What is the equation that relates  $G$  to  $K$ ?:  $\Delta G = -RT \ln K$
- What is the Boltzmann constant?:  $\ln(\# \text{ of microstates})$
- What does a big  $W$  mean?: More microstates

- What does more microstates mean?: Big  $\Delta S$ - more entropy

- What does fewer microstates mean?: Smaller  $\Delta S$ - less entropy

- What happens to entropy when volume is increased?: Microstates increase, entropy increases

- What is the relationship between  $S$

of a system,  $q$ , and  $T$ ?:  $\Delta S$

$$(\text{system}) = q_{\text{rev}}/T$$

- What happens to entropy  $S$  when temperature increases?: It increases

- What is the relationship between mass and entropy  $S$ ?: As mass increases, entropy increases

- What is the relationship between molecular complexity and entropy  $S$ ?: As molecular complexity increases, entropy increases

- What is the relationship between  $G$  and  $Q$ ?:  $\Delta G = RT \ln Q$

# Thermochemistry Study Guide Answers

**Julia Schneider**



## **Thermochemistry Study Guide Answers:**

Thermochemistry Study Guide Answers: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the captivating narratives that have captivated audiences this year. The Must-Read : Colleen Hoover "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Thermochemistry Study Guide Answers : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Thermochemistry Study Guide Answers : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Popen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and suspenseful novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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