Solubility and Thermodynamics

DETERMINING THE K_{sp} OF Ca(OH)₂ AT ROOM TEMPERATURE

	5°C 0172M		
Volume of HCl used Moles of OH	1.60×10-4 mol	9.40 mL	9-32-mL 1-60x10-4
Volume of sample [OH ⁻]	5.00	5.00	5-00
[Ca ²⁺] K _{sp}	1 PA 102 B = 1	Trioxio 1	O.G. WAL
Experimentally determined value of K_{sp} for trial 1	PSP NO 5	±	standard deviation
A	2 6	THE SOURCE THE	

Thermodynamics Lab Answers

G Thomas

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