

at about constant pressure. This cycle later formed the basis for modern gas turbine plants.

Gottlieb Daimler (1834–1900) obtained a patent in 1879 for a multicylinder automotive engine, which was commercially successful. Dr. Rudolf Christian Karl Diesel (1858–1913) studied at Technische Hochschule in Munich. He designed large steam engines and boilers. He later developed in 1897 an I.C. engine with fuel injection which resembled the modern diesel engine. Failing health, continuing criticism and serious financial setbacks beset Diesel who in 1913 disappeared from a boat crossing the English channel in a moonlit night. Josiah Willard Gibbs (1839–1903) is often regarded as the most brilliant thermodynamicist produced in the USA. He received the first doctorate degree in engineering in the USA (Yale University). He contributed significantly in many areas of thermodynamics like heterogeneous systems, phase rule, physical chemistry and statistical thermodynamics. Some of his very important papers were published in obscure journals like Connecticut Academy of Sciences, and remained unknown to most scientists. Only after his death, these were discovered.

### SOLVED EXAMPLES

**Example 1.1** The pressure of gas in a pipe line is measured with a mercury manometer having one limb open to the atmosphere (Fig. Ex. 1.1). If the difference in the height of mercury in the two limbs is 562 mm, calculate the gas pressure. The barometer reads 761 mm Hg, the acceleration due to gravity is  $9.79 \text{ m/s}^2$ , and the density of mercury is  $13,640 \text{ kg/m}^3$ .

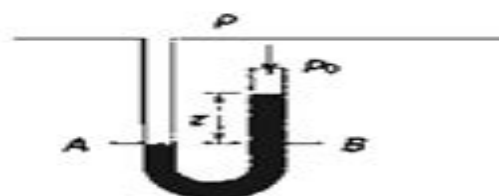


Fig. Ex. 1.1

**Solution** At the plane  $AB$ , we have

$$p = p_0 + \rho g z$$

Now

$$p_0 = \rho g z_0$$

where  $z_0$  is the barometric height,  $\rho$  the density of mercury and  $p_0$  the atmospheric pressure.

Therefore

$$\begin{aligned} p &= \rho g (z + z_0) \\ &= 13,640 \text{ kg/m}^3 \times 9.79 \text{ m/s}^2 (0.562 + 0.761) \text{ m} \end{aligned}$$

# Theory Of Machine By Pk Nag

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