EXPERIMENT 2

ELECTROCHEMISTRY: ELECTROCHEMICAL CELL AND THERMODYNAMIC FUNCTIONS

Objectives

- 1. To construct electrochemical cells
- To determine the net cell potentials for three electrochemical cells
- To calculate the net cell potential based on the potentials of the half-reactions that occur and to compare your experimental and calculated values.
- To measure the Gibbs energy based on the cell potentials.

Introduction

The use of electrochemical cells to convert the Gibbs energy stored in the constituent half-reactions into electrical work is of enormous industrial as well as fundamental significance. We have all used batteries and these are simply galvanic cells that are similar to those you will be constructing in this experiment. In the laboratory, a typical electrochemical cell has the following general construction:

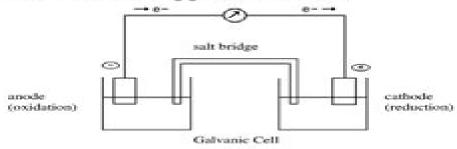


Figure 2.1: A schematic diagram of galvanic cell

The standard reduction potential is a quantitative measure of a substance's tendency to accept electrons under standard conditions (1 atm of gases and 1 M concentration) and is measured in volts. The more positive the reduction potential, the greater the tendency for reduction to occur. For the half-reaction that represents the oxidation of a metal, you can determine the standard oxidation potential (E^*_{ox}) for the half-reaction. This value has the same numerical value as the standard reduction potential but the opposite sign. For example, aluminum ion has a standard reduction potential value of $-1.65 \, \text{V}$

$$Al^{3*}(aq) + 3e \rightarrow Al(s)$$
 $E^{\circ}_{red} = -1.65 \text{ V}$

Thermodynamics Lab Manual Experiment

J. A. Beran, Mark Lassiter

Thermodynamics Lab Manual Experiment:

Experiments in Heat Transfer and Thermodynamics Robert Alan Granger, 1994-06-24 Engineering curricula are notoriously demanding One way to make the material easier to grasp and more fun to learn is to emphasize the experimental or hands on aspects of engineering problems. This unique book is about learning through active participation in laboratory experiments and it specifically aims to dispel some of the mystery so many students associate with the study of thermodynamics and heat transfer In it the author presents a collection of experiments in heat transfer and thermodynamics contributed by leading engineering educators The experiments have been tested evaluated and proved successful for classroom use Each experiment follows the same step by step format which includes the objective of the experiment apparatus needed procedure suggested headings and references The experiments use apparatus that is easily built or attainable Among the topics covered are heat conduction convection boiling mixing diffusion radiation heat pipes and exchangers and thermodynamics The book will be especially useful as a companion to standard heat transfer and thermodynamics texts Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don t write experiments A student s scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment s Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section Lab-At-Home Experiments in Chemistry and Chemical Engineering Jude A. Okolie, Ugochukwu

Patrick Okoye, 2025-10-28 With the rise of remote and hybrid learning there is an increasing need for flexibility around laboratory based coursework This book offers readers a guide to conducting a wide variety of chemistry and chemical engineering labs and experimental procedures that can be completed easily and safely outside of a traditional lab setting including at home It helps students and interested readers achieve hands on learning of chemistry and chemical engineering based concepts without the need for sophisticated lab equipment Features a comprehensive range of labs on such topics as separation processes CO2 capture 3D printing reaction kinetics and fuel cells Includes detailed information on how to perform experiments at home via virtual experimental procedures and accompanying videos Uses household and readily accessible materials and provides safe material handling and disposal methods Offers solution spreadsheets where experimental results can be computed and validated Gives background information and relevant theory about each experiment and includes exercises at the end of each chapter Provides Python code and instructions on how to perform experiments including a bowling example This practical and useful guide is aimed at chemistry and chemical engineering students as well as general readers interested in running experiments in these disciplines Video supplements to support comprehension of each lab and a solutions spreadsheet for Chapter 1 are available for download ChemDiscovery Lab Manual Olga I. Agapova, 2002 Cumulative Book Index ,1925 A world list of books in the English language The United States Catalog ,1925 Designing and Writing the Thermodynamics Lab Manual Carol A. Reed, 1988 Laboratory Manual [in] Engineering Physics ... Seville Chapman, 1947 <u>Laboratory Manual/Human Nutr 2</u> Dashmanb, 1996-04-01 Utilization of the laboratory for nutrition support accompanies the greater demand for quality nutrition as evidenced by the recent nutrition label law Because quality nutrition is also good preventive medicine nutrition assessment may be part of a preliminary examination This book introduces several areas of nutrition research that the American Institute of Nutrition recently detailed these include animal nutrition diet and disease energy and macronutrient metabolism growth and development neuroscience nutrient gene interactions nutrient and food toxicity public health nutrition policy and vitamins and minerals The experiments in this laboratory manual provide the basics of nutritional assessment including anthropometric biochemical clinical dietary and environmental parameters Biological food processing food composition theoretical principles and the effect of pharmaceuticals on appetite absorption metabolism and behavior are also studied

Teaching Science Online Dietmar Kennepohl,2023-07-03 With the increasing focus on science education growing attention is being paid to how science is taught Educators in science and science related disciplines are recognizing that distance delivery opens up new opportunities for delivering information providing interactivity collaborative opportunities and feedback as well as for increasing access for students This book presents the guidance of expert science educators from the US and from around the globe They describe key concepts delivery modes and emerging technologies and offer models of practice The book places particular emphasis on experimentation lab and field work as they are fundamentally part of the

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American Journal of Physics, 1993

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