

Solution Manual for Fundamentals of Chemical Engineering Thermodynamics 1st Edition by Dahm

Chapter 5: Thermodynamic Processes and Cycles

5-6) This problem examines the Rankine heat engine introduced in Figure 5-5. Saturated steam at $T = 250^\circ\text{C}$ enters the turbine and the condenser operates at $T = 40^\circ\text{C}$.

- A) Assuming the turbine is reversible, give your best estimate of the efficiency of the cycle, and indicate the quality of the stream leaving the turbine.
- B) Assuming the turbine has an efficiency of 75%, give your best estimate of the efficiency of the cycle, and indicate the quality of the stream leaving the turbine.
- C) Find the flow rate of circulating water needed to produce a net power of 1 MW from the cycle with the turbine efficiency of 75%.

Set an entropy balance around the adiabatic reversible turbine **Solution:**

$$\frac{d(M\hat{S})}{dt} = \dot{m}_{in}\hat{S}_{in} - \dot{m}_{out}\hat{S}_{out} + \frac{\dot{Q}}{T} + \dot{S}_{gen}$$

Cancelling terms

$$\hat{S}_{in} = \hat{S}_{out}$$

Find \hat{S}_{out}

$$(\hat{S}_{in}) \rightarrow \text{Saturated Water vapor at } 250^\circ\text{C} \rightarrow 6.072 \frac{\text{kJ}}{\text{kg K}}$$

$$\hat{S}_{out} = 6.072 \frac{\text{kJ}}{\text{kg K}}$$

Find quality of mixture leaving reversible turbine:

$$\hat{S}_{out} = 6.072 \frac{\text{kJ}}{\text{kg K}} = (1 - q)\hat{S}_L + q\hat{S}_V$$

At 40°C :

$$\hat{S}_L = 0.572 \frac{\text{kJ}}{\text{kg K}}; \hat{S}_V = 8.256 \frac{\text{kJ}}{\text{kg K}}$$

$$q = 0.716$$

Thermodynamics Solution Manual Chapter 6

Hans U. Fuchs



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Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications *Student's Solutions Manual for Thermodynamics, Statistical Thermodynamics, and Kinetics* Thomas Engel, Philip J. Reid, 2006 *Student Solutions Manual for Tipler and Mosca's Physics for Scientists and Engineers, Sixth Edition: Chapters 1-20* Todd Ruskell, 2008

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